CAMBRIAN COLLEGE of Applied Arts and Technology Sault Ste. Marie

COURSE OUTLINE

METALLURGY

MET 201-4

Note: This course is offered as an option and may be taken.instead of the "Pulp & Paper Technology" CHM 201-4 in the Industrial Chemistry Technician Program. MET 201-4

TEXT:

Elementary Metallurgy & Metallography by Schrager

REFERENCE:

Physical Metal	ysical Metallurgy		er	
Metallurgy		- Johnson & Weeks		
Handbook of No	onferrous	Metallurgy	- Liddel	
Handbook of Nonferrous Metallurgy				
(Recovery of	Metals)	- Lidd	ell	

Metallurgy

Option - (Industrial Chemistry)

MET 201-4

Topic Number Periods

Topic Information

Iron & Steel

a) <u>Ores</u>

Definition Types Location of deposits

b) Blast Furnace Process

History Redaction Process Slag Uses

c) Cast Iron

Definition Types - Gray White Ductile ^ Malleable Wrought

Uses

d) Steelmaking Processes

Definition History Open Hearth Process Basic Oxygen Process Electric Arc Process

Copper and Zinc

a) <u>Ores</u>

Location of deposits Extractive processes

b) Properties 6c U^{se}s

Strength Ductility Modulus of elasticity Fatique Corrosion resistance

Metallurgy

Option - Industrial Chemistry

MET 201-4

Topic Number	Periods	Topic Information
		c) Fabrication of Copper & its Alloys
		Casting Hot & cold working Electroplating Joining
		d) <u>Alloys</u>
		Types Properties Uses
3	3	Aluminium
		a) <u>Ores</u> Location of deposits Extractive process
		b) <u>Properties & Uses</u>
		Corrosion resistance Fatique strength Creep strength Mechanical & physical Heat & electrical conductivity Weight
		c) <u>Fabrication</u>
		Casting Mechanical working Joining
		d) <u>Alloys</u>
		Types Properties Uses
4	3	Magnesium
		a) <u>Ores</u>
		Location of deposits Extractive process
		k) <i>E]i2P</i> A <u>rties &</u> U <u>ses</u>

Wear resistance

Metallurgy

Option - Industrial Chemistry

MET 201-4

Topic	Number	Periods	Top	ic Information
				Properties & Uses cont'd.
				Fatique strength Creep strength Corrosion resistance Weight
			C)	Fabrication
				Casting Mechanical working Joining
			d)	Alloys
				Types Properties Uses
				Nickel
			a)	Ores
				Location of deposits Extractive processes
			b)	Properties & Uses
			C)	Effects of Alloying Elements
			d)	Nickel Bases Alloys
				Types - Monels Ni-Si-Cu base Inconel Ni-Fe base Ni-Mo-Fe base Properties Uses
				Lead
			a)	<u>Ores</u> Location of deposits Extractive processes
			b)	Properties & Uses

c) Alloys

Metallurgy Option - Industrial Chemistry MET 201-4

Topic Number	Periods		Topic Information
7	1	Tin	
		a)	Ores
			Location of deposits Extractive processes
		b)	Properties & Uses
		c) 1	Alloys
8	30	Labo	pratory
		a)]	Hardness Tests
]	File Brinell Rockwell Scleroscope
		b) I	Microscopy
		:]]]	Sampling Polishing Mounting Microetching Macroetching Metallurgical microscope
		c) <u>1</u>	X-ray
			Gamma ray Magnetic testing
		d) I	Heat Treatment
			Annealing Normalizing Induction hardening Case hardening Quench hardening Cyanide hardening

The student should be able to:

- 1. Give a definition of Metallurgy and outline the subjects dealt with in Extractive, Physical, and Mechanical Metallurgy.
- 2. Give a definition of Mineral dressing.
- 3. Describe methods used to liberate minerals before separation, such as Crushing, Crinding, Autogenous milling.
- A- To give a definition of the degree of liberation,
- 5. To discuss and calculate the recovery of metals and the problem of middlings.
- 6- Describe Methods of Magnetic Separation and name types of ores separated this way.
- 7. Describe methods of Heavy medid separation and Gravity separation.
- 8. Describe the principle of Flotation and to perform a separation on a Laboratory Flotation cell.
- 9. Describe the process of Sintering.
- 10. Describe the process of Pelletizing.
- 11. Describe the process of coke making.
- 12. Name the raw materials used for iron makings and-explain their functions.
- 13. Describe the operation of a Blastfurnace.
- 14. Describe the chemical reactions taking place in the Blast furnace.
- 15. Give the material balance of a blast furnace.
- 16. Give the composition of Pig Iron.
- 17. Describe processes of steel making, the Puddling process, the Bessemer Process, the Open Hearth Process and the Basic Oxygen process.
- 18. Describe the relationship between Carbon Content, melting point of steel and components in steel. (Iron-Carbon Diagram)
- 19. To explain on the iron carbon diagram the meaning of the critical temperature lines, the culectic-and the eutectoit point.
- 20. Be able to identify under the microscrop the components of steel, Ferrite, Pearlite, Martinsite and explain their composition.
- 21. Be able to prepare steel specimens for microscopic examination by grinding, polishing and etching.
- 22. Be able to determine the hardness of steel by using the Rockwell hardness tester and the Shoreslerscope.
- 23. To explain other methods of hardness determinations, the Brinell and Vickers system.
- 24. To explain the following heat treatment processes: Annealing, Normalizing and Tempering.

To explain the following methods of Case hardening: Cyaniding, Pack carburizing, Nitriding, Flame hardening.

To discuss the relationship between hardness of steel, heat treating temperature and Carbon content from date obtained from experiments.

To discuss the effect of Quenching media on the hardness of steel from data obtained from experiments.